# SELF-ORGANIZATION AND DISSIPATIVE STRUCTURE FORMATION IN NON-CRYSTALLINE MATERIALS

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A model is proposed to describe dissipative structures formation in non-crystalline materials. To investigate temperature dependencies of the share of atoms in liquid-like states and mean-square atomic displacements. It is shown that system transits into non-equilibrium state when the melt freezing rate increases which corresponding of dissipative structures formation in non-crystalline materials .

#### Introduction

Non-crystalline state of matter represents peculiar self-organizing system with certain degree of ordering and life time, which are determined by external control parameters (temperature, pressure, electromagnetic radiation), and includes sufficiently wide range of materials from glassy matters with organic and non-organic origin to bio-object [1,2,3]. It provides broad range of their properties and character of interaction with external fields.

Example of self-organizing systems is dissipative structures formation, when freezing the melt, for which all above enumerated requirements are fulfilled (technology of freezing process causes presence of open system; feedback is provided by interaction of field of dynamic displacements, viscosity and force constants; transition realizes under certain rate of external conditions change). Solution of this problem is connected with investigation of structure and properties of equilibrium (crystalline, quasicrystalline, liquid) and strongly non-equilibrium (non-crystalline) materials in the vicinity of phase transition.

#### Model of system

Crystallization of the melt takes place during equilibrium liquid-crystal phase transition, as temperature lowers under thermodynamic equilibrium. Transition model must take into account anharmonicity of atomic oscillations, what is provided by use of self-consistent phonons theory (SCPT) [4] so long as classic harmonic and quasi-harmonic approximations for system in transforming temperature region are not applicable. When heating, as amplitude of heat oscillations increases, quantity of defects goes up, what

leads to their uniting. In phase transition vicinity concentration of such fluctuation essentially rises and as homogeneous atomic heat oscillations has considerable influence on the physical properties of the system [5].

As we approach the temperature T of the solid to the melting temperature T<sub>m</sub>, the concentration of the point defects (the internodal atom (vacancy) type) or the extended defects (the dislocation type) increases sufficiently. Such defects taken separately are not the disordered phase nuclei. However, if they are accumulated within a certain macroscopic domain, then such area of disordering corresponds to the structure inherent to the liquid-like (soft) states, i.e. it has a shearing instability, a high level of dynamic displacements and the instability related to the defect formation. Hence, within the phase transformation interval along with the homogeneous fluctuations (i.e. the thermal displacements

The soft states are the specific structural states with the additional degree of freedom in the configurational space. These states are due to the spatial fluctuations of the short-range order parameters (the number and the length of the bonds, the interbond angles) around their average value and the formation of the defect states in the highly non-equilibrium conditions and are intrinsic in the non-crystalline solids. Atoms in the soft states process a considerable static displacements level of consequently, the vibrational anharmonism and the lability to the spatial rearrangements. The bonding topology and the configurational parameters which describe the soft states are accounted for and detailed in terms of certain structural models. We shall analyse the general tendencies of the formation of macroscopically

ordered structures in non-crystalline solids away from the equilibrium state, therefore, we shall use the universal characteristics of the soft states.

Consider the system of N similar atoms of mass M, the portion  $N_1$  of which being in

$$H = \sum_{l=1}^{N} \sum_{j} T_{f}(l) \ \sigma_{f}(l) + \frac{1}{2} \sum_{l \neq l'} \sum_{f, f'} \Phi_{ff'}(\vec{r}_{ll'}) \ \sigma_{f}(l) \ \sigma_{f'}(l'),$$

where  $T_f(l) = \frac{\vec{P}^2(l)}{2M}$  is the kinetic energy of the atom l,  $\Phi_{ff'}(\vec{r}_{ll'})$  is an interatomic interaction potential and  $\sigma_f(l)$  characterises two sets of locally observed microscopic states:

the solid-like state and the portion  $N_1 = N - N_1$  being in the liquid-like state. The Hamiltonian of the system can be presented within the binary atom interaction approximation in a following form:

 $\sigma_f(l) = \begin{cases} I, & \text{if the atom } l \text{ is in the state } f; \\ 0, & \text{if the atom } l \text{ is not in the state } f. \end{cases}$ 

The possibility of the formation of the liquid-like states in the solid is, consequently,  $\sigma_f = \langle \sigma_f(t) \rangle = \sqrt[N_f]{}_N$ . Let us introduce the effective Hamiltonian of the system:

$$H_{0} = \sum_{f} H_{0f} + (1 - \sigma_{2})^{2} \sum_{l,l',\alpha} I_{\alpha}(l,l') v_{\alpha}(l), \quad v_{\alpha}(l) = \sum_{m} v_{\alpha} \left( \frac{l}{m} \right) \sigma_{2}(m),$$

$$H_{0f} = \sigma_{f} \sum_{l} T_{f}(l) + \frac{\sigma_{f}^{2}}{2} \sum_{l,l',\alpha} \sum_{\alpha, \beta} \left[ \Phi(\bar{R}_{ll'}) + \frac{1}{2} u^{\alpha}(l,l') \tilde{\Phi}_{f}^{\alpha\beta}(l,l') u^{\beta}(l,l') \right],$$
(2)

whose model parameters, i.e. the fraction of atoms in the liquid-like states  $\sigma_f$ , the force constants  $\tilde{\Phi}_f^{\alpha\beta}(l,l')$  and the mean-square relative displacements of atoms

 $y_{\alpha\beta}(l,l') = \langle u^{\alpha}(l,l') u^{\beta}(l,l') \rangle$  are calculated on the basis of the variational principle for the free energy functional:

$$F \le F_{\theta} + \langle H - H_{\theta} \rangle_{\theta}, \quad F_{\theta} = \sum_{f} F_{\theta f} - TS, \quad F_{\theta f} = -\Theta \ln \left\{ SPe^{\frac{-H_{\theta f}}{\Theta}} \right\},$$
 (3)

Here 
$$S = -k_b \ln \left[ \prod_{f} \left\{ \frac{g_f!}{N_f! (g_f - N_f)!} \right\} \right]$$
 is the

entropy,  $g_f$  is a statistical weight of the state f,  $r_l = R + u_l + v_l$ ,  $u_l$  and  $v_l$  are the dynamic and the static displacements of atoms,  $u_{\alpha}(l,l') = u_{\alpha}(l) - u_{\alpha}(l')$  are the relative

displacements,  $v_{\alpha}(l/m)$  are the static displacements of the atom l when the atom m is in the liquid-like state,  $\Theta = k_b T$ . The variation of F over the parameters  $\sigma(\sigma = \sigma_2)$ ,  $\widetilde{\Phi}_f^{\alpha\beta}(l,l')$ ,  $y_{\alpha\beta}(l,l')$  at fixed temperature T and pressure P:

$$\delta F = \left[\frac{\partial F}{\partial \sigma}\right]_{y_{\alpha\beta},\tilde{\Phi}_{f}^{\alpha\beta}} \cdot \delta \sigma + \left[\frac{\partial F}{\partial \tilde{\Phi}_{f}^{\alpha\beta}}\right]_{\sigma,y_{\alpha\beta}} \cdot \delta \tilde{\Phi}_{f}^{\alpha\beta} + \left[\frac{\partial F}{\partial y_{\alpha\beta}}\right]_{\sigma,\tilde{\Phi}_{f}^{\alpha\beta}} \cdot \delta y_{\alpha\beta} \tag{4}$$

allows one to determine their temperature dependence for the equilibrium  $(\delta F = 0)$  and non-equilibrium  $(\delta F \neq 0)$  transformations. Let us find the functional equations to determine

the probability of the formation of the liquidlike states, force constants and mean-square displacement of atoms. Using equations (2) and (3) for  $\sigma$  we find:

$$\frac{\partial F}{\partial \sigma} = \frac{\partial F}{\partial \sigma} + \frac{\partial}{\partial \sigma} \left\langle H - H_0 \right\rangle_0 = \tilde{T}_2 - \tilde{T}_1 + 2\sigma \left( \tilde{\Phi}_2 - \tilde{\Phi}_1 \right) - 2\tilde{\Phi}_1 - 4\tilde{I} \left( 1 - \sigma \right) \sigma - 2IN \ln \frac{g_2}{\sigma} - 1. \tag{5}$$

Here the following notation has been introduced:  $\tilde{T}_f = \sum_l \left\langle \frac{\tilde{P}_f^2(l)}{2M} \right\rangle$ ,

$$\widetilde{\Phi}_f \approx \frac{1}{2} \sum_{l,l'} \left\langle \Phi_f \left( l, l' \right) \right\rangle, \qquad \widetilde{I} = \left\langle \sum_{l,l'} \sum_{\alpha,m} I_{\alpha} \left( \overrightarrow{r}_{ll'} \right) v_{\alpha} \left( \frac{l}{m} \right) \right\rangle.$$

Similarly, the functional equations with

respect to the quantities  $y_{\alpha\beta}(l,l')$  and  $\tilde{\Phi}_f^{\alpha\beta}(l,l')$  are found:

$$\frac{\partial F}{\partial y_{\alpha\beta}(l,l')} = \frac{\partial}{\partial y_{\alpha\beta}(l,l')} \left\langle H - H_{\theta} \right\rangle_{\theta} = \frac{1}{4} \left( \left\langle \nabla^{\alpha}_{ll'} \nabla^{\beta}_{ll'} \Phi_{f}(\hat{r}_{ll'}) \right\rangle \sigma_{f}^{2} - \widetilde{\Phi}_{f}^{\alpha\beta}(l,l') \right), \tag{6}$$

$$\frac{\partial F}{\partial \widetilde{\Phi}_{f}^{\alpha\beta}(l,l')} = \frac{\partial F_{0}}{\partial \widetilde{\Phi}_{f}^{\alpha\beta}(l,l')} + \frac{\partial}{\partial \widetilde{\Phi}_{f}^{\alpha\beta}(l,l')} \left\langle H - H_{0} \right\rangle_{0} = \frac{\hbar}{2 \, \text{MN}} \sum_{\vec{k}} \frac{\sin^{2} \frac{\vec{k} \vec{r}_{ll'}}{2}}{\omega(\vec{k})} \coth \frac{\hbar \omega}{2\Theta} - \frac{y_{\alpha\beta}(l,l')}{4} \,. \tag{7}$$

In the non-equilibrium non-crystalline state,  $(\delta F \neq 0)$ . Expanding F into a power series in the system deviation from the equilibrium state and restricting ourselves to the first terms of the expansion, we have:

$$F = F_0 + \frac{1}{2} \sum_{i,j} a_{ij} \eta_i \eta_j + \frac{c}{3} \eta^3 + \frac{b}{4} \eta^4,$$
 (8)

where the following notations are used:

$$\begin{split} \eta &= \sigma - \sigma_e, \eta_y = \left\{ \eta_y^{\alpha} \right\}, \ \eta_y^{\alpha} = y_{\alpha}(l, l') - y_{\alpha}(l, l')_e, \\ \eta_{\phi} &= \left\{ \eta_{\phi}^{\alpha} \right\}, \ \eta_{\phi}^{\alpha} = \widetilde{\Phi}^{\alpha\alpha}(l, l') - \widetilde{\Phi}^{\alpha\alpha}(l, l')_e, \\ a_{ij} &= \left( \frac{\partial^2 F}{\partial \eta_i \partial \eta_i} \right), \ c_{ijk} = \frac{l}{2} \left( \frac{\partial^3 F}{\partial \eta^3} \right), \ b_{ijk} = \frac{l}{6} \left( \frac{\partial^4 F}{\partial \eta^4} \right) \end{split}$$

and the allowance was made for  $\left(\frac{\partial F}{\partial \eta_i}\right)_e = 0$ 

(the subscript "e" corresponds to the equilibrium state).

To describe the kinetics of the order parameter variation we shall use the Landau-Khalatnikov regression equation [7]  $\frac{\partial \eta_i}{\partial t} = -\gamma_i \left( \frac{\partial F}{\partial \eta_i} \right)$  which allows one in relations

(8) to obtain the following system of nonlinear kinetic equations:

$$\lambda^{-1} \frac{\partial \eta}{\partial t} = -a_{11} \eta - a_{12} \eta_y - a_{13} \eta_{\Phi} - c \eta^2 - b \eta^3 - D_{\eta} \nabla \eta,$$

$$\lambda^{-1} \frac{\partial \eta_y}{\partial t} = -a_{12} \eta - a_{22} \eta_y, \quad \lambda^{-1} \frac{\partial \eta_{\Phi}}{\partial t} = -a_{13} \eta - a_{33} \eta_{\Phi}.$$
(9)

The first terms in these equations describe the energy dissipation with relaxation time of  $\tau_{ij} = a_{ij}^{-1}$ , while the second terms define the collective mode interaction.

Let us draw on the Prigogine principle of mode subordination which holds for highly non-equilibrium systems [8]. According to this principle:

$$\frac{\partial \eta_y}{\partial t} = 0, \ \frac{\partial \eta_{\Phi}}{\partial t} = 0.$$

This allows one to reduce the number of order parameters:  $\eta_y = -\left(\frac{a_{12}}{a_{22}}\right)\eta$ ,  $\eta_{\phi} = -\left(\frac{a_{13}}{a_{33}}\right)\eta$  and rewrite (9) in the following form:

$$\lambda^{-1} \frac{\partial \eta}{\partial t} = -a_0 \eta - c \eta^2 - b \eta^3 - D_{\eta} \nabla \eta, \ a_0 = -\left(a_{II}^2 - \frac{a_{I2}}{a_{22}^2} - \frac{a_{I3}}{a_{33}^2}\right). \tag{10}$$

The expansion coefficients are the functions of the temperature, pressure and control parameter, i.e. the cooling rate is:

 $a_0 = a_0(T, P, q)$ . Since in the equilibrium state:  $\delta F = 0$ ,  $\delta^2 F > 0$  and  $\eta = 0$ , then, to ensure this,  $a_0(T, P, q)$  must be a positively defined matrix,

i.e.  $a_0(T, P, q) > 0$ . In the non-equilibrium state  $\delta F \neq 0$ ,  $\eta \neq 0$ . To provide the formation of the stationary states with non-zero parameter, it is essential that  $a_0(T, P, q) < 0$ . Thus, at the transition to the non-equilibrium non-zero order  $\eta = \frac{-2c \pm \sqrt{4c^2 - 4a_0b}}{6b}, \quad \text{and} \quad \text{the}$ expansion coefficient is  $a_0(T_c, P_c, q_c) = 0$ . following conditions  $c(T_c, P_c, q_c) = 0$ ,  $b(T_c, P_c, q_c) > 0$ , since, with the approach to the point  $T_c, P_c, q_c$  from the equilibrium state side, an even derivative should be positive. Thus, in the process of melt cooling and at the transition to the non-

crystalline state, 
$$a_0(T_e, P_e, q_e)$$
 can be approximated by:

$$a_{\theta}(T_c, P_c, q_c) = -\tilde{a}_{\theta} \ln(\arctan[\tilde{q}]), \ \tilde{q} = \frac{q - q_c}{q_c},$$

which is transformed to  $a_0(T_c, P_c, q_c) \approx -a_0 \tilde{q}$  as  $\tilde{q} \ll 0$ .

Thus, started from expressions (5)-(10), we arrive at the following self-consistent system of equations with respect to the relative fraction of atoms s in the liquid-like states, the mean-square atomic displacements  $y_{\alpha\beta}(l,l')$  and force constants  $\tilde{\Phi}_{\alpha\beta}^{\alpha\beta}(l,l')$ :

$$F_1(\sigma) = 0, F_2(y) = 0, F_3(\Phi) = 0$$
. (11)

Here:

$$F_{I}(\sigma) = \widetilde{T}_{2} - \widetilde{T}_{I} + 2\sigma(\widetilde{\Phi}_{2} - \widetilde{\Phi}_{I}) - 2\widetilde{\Phi}_{I} - 4\widetilde{I}(I - \sigma)\sigma - 2\Theta N \ln \frac{g_{2}}{\sigma} + \widetilde{a}_{0}\widetilde{q} \eta - c\eta^{2} - b\eta^{3} - D_{\eta}\nabla^{2}\eta,$$

$$\begin{split} F_2(y) &= \frac{\hbar}{2\,MN} \sum_{\vec{k}} \frac{\sin^2\frac{\vec{k}\vec{r}_{B'}}{2}}{\omega(\vec{k})} \coth\frac{\hbar\omega}{2\Theta} - \frac{y_{\alpha\beta}(l,l')}{4} - \frac{\eta_y}{\tau_y}, \\ F_3(\Phi) &= \frac{1}{4} \Big( \Big\langle \nabla^{\alpha}_{B'} \nabla^{\beta}_{ll'} \Phi_f(\vec{r}_{B'}) \Big\rangle \sigma_f^2 - \widetilde{\Phi}_f^{\alpha\beta}(l,l') \Big) - \frac{\eta_\Phi}{\tau_\Phi}. \end{split}$$

This system allows the temperature properties of the system to be studied both at the equilibrium  $(\eta = 0)$  and highly non-equilibrium  $(\eta \neq 0)$  transformations.

## Self-consistent description of nonequilibrium systems

Consider the application of the above concepts to the studies of the stability and metastable states formation in the certain solid-state model in the high-temperature region ( $\theta >> \hbar\omega_L$ ,  $\omega_L$  is a maximum frequency of vibrations). Let us describe the interatomic interaction through the effective pair potential of the following type  $\Phi(r) = \Phi_1(r) + \Phi_2(r)$ , where its short-range component  $\Phi_2(r)$  can be approximated by the hard-sphere potential ( $\Phi_2(r) = \Phi_0$  at r < a or  $\Phi_2(r) = 0$  at r > a) and the long-range one  $\Phi_I(r)$  can be presented in the form of a sum of the central Morse-like

$$\left( \Psi(r) = V_0 \left[ exp \left\{ -12 \frac{r-a}{r} \right\} - 2 exp \left\{ -6 \frac{r-a}{r} \right\} \right] \right) \text{ and}$$

the non-central potential 
$$\left(G(\delta) = -G_0 \exp\left\{-4\left(\frac{\delta}{a}\right)^2\right\}\right), \text{ where } \delta \text{ is the}$$

deflection of the atom in the plane normal to the direction of the bond,  $G_0$  and  $V_0$  are the potential parameters.

Solition results of self-consistent system (12) in equilibrium and non-equilibrium states with the use of the iteraction procedure are shown in Fig.1. In temperature dependencies ( $\tau$ = $\theta$ / $V_0$  is a reduced temperature) of share of atoms in liquid-like states we can mark out five distinctive parts which correspond to different system states:

- stable: part a-b crystalline ordered state, part e-f - liquid disordered state;
- metastable: part b-c super-heated crystal, part d-c - super-cooled liquid;
- virtual system state: part c-d state is thermodynamically unstable.

During increase of the melt freezing rate, the narrowing of temperature range of metastable regions existence and phase

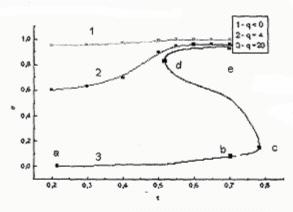


Fig. 1. Temperature dependence of the share of atoms in liquid-like states.

transition point shift to lower temperatures area are observed. When q reaches certain critic value metastable regions completely vanish and system transits into non-equilibrium state. The degree of the system deviation from equilibrium is defined by deviation of order parameter from its equilibrium value (Fig.1).

Non-crystalline materials formation needs considerable system deviation from equilibrium state when phase transition takes place. Deviation value is defined by external control parameter-freezing rate. As freezing rate is higher than critic to represent transition character we must take into account occurred in the system bifurcations, which lead to dissipative structures formation. For example, in region q <q c, system is in equilibrium under different values of external control parameter [10].

During the melt cooling under strongly non-equilibrium conditions the part of system energy, which is in the atoms and molecules heat motion, transits in macroscopic organized motion by spatial-temporal correlation. This motion stipulates formation of complex spatial system organization (dissipative structure), when the largest dissipation of the energy is observed. In stationary case, formed dissipative structures are stable formations with distinctive shape and volume, and as shape so volume is stable to low disturbances.

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## САМООРГАНІЗАЦІЯ ТА ФОРМУВАННЯ ДИСИПАТИВНИХ СТРУКТУР В НЕКРИСТАЛІЧНИХ МАТЕРІАЛАХ

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Розглянуто модель формування дисипативних структур в некристалічних матеріалах. Досліджено температурну залежність долі атомів в рідкоподібних станах, середньоквадратичних зміщень атомів. Показано, що в залежності від швидкості охолодження відбувається перехід системи в нерівноважний стан, який супроводжується формуванням дисипативних структур.